

12th Science: Chemistry Amines,

| | DATE: |
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| | TIME: 1 hour |
| | MARKS: 25 |
| SEAT NO: | |

Note:-

- 1. All Questions are compulsory.
- 2. Numbers on the right indicate full marks.

Section A

Q.1 Select and write the correct answer.

(4)

- The weakest base among the following is
 - A) Dimethylamine
- B) aniline
- C) methylamine
- D) ethylamine
- 2. Isobutylamine is an example of
 - A) 2° Amine
- B) 3° Amine
- C) 1° Amine
- D) quaternary ammonium salt
- 3. The strongest base among the following is
 - A) $C_6H_5NH_2$ B) $(C_6H_5)NH$

 - C) NH_3 D) $(C_6H_5)_2NH$
- Aniline when diazotized in cold and then treated with dimethyl aniline gives a coloured product its structure would be
 - A) $CH_3 \bigcirc -N = N \bigcirc -NH_2$ B) $(CH_3)_2 N \bigcirc -N = N \bigcirc -N$
 - C) $(CH_3)_2 N \bigcirc -NH \bigcirc \bigcirc -NHCH_3$

Q.2 Answer the following.

(3)

(2)

- 1. What is the formula of nitrous acid? Can nitrous acid be stored in bottle?
- 2. Can tertiary amine undergoes acetylation reaction? Explain.
- 3. Arrange the following:
 - (a) A decreasing order of Boiling Point : $C_2H_5 OH$, $C_2H_5 NH_2$, $(CH_3)_2NH$
 - (b) Increasing order of solubility : $C_2H_5 NH_2$, $C_3H_7 NH_2$, $C_6H_5 NH_2$

Section B Attempt any Four

Q.3 Complete the following reaction:

$$C_6H_5N_2^+Cl^2 + C_2H_5OH \longrightarrow ?$$
 (b) $C_6H_5NH_2 + Br_{2(aq)} \longrightarrow ?$

(a)
$$C_6H_5N_2^{\dagger}Cl^{-} + C_2H_5OH \longrightarrow 9$$

| Q.4 | Write chemical equation for : (a) Reaction of alc. NH_3 with C_2H_5I (b) Ammonolysis of benzyl chloride followed by the | (2) | |
|---------------------------|---|-----|--|
| | reaction with two moles of CH ₃ – I | | |
| Q.5 | why pk _b Value of diethyl amine is less than that of ethyl amine? | (2) | |
| Q.6 | Ethyl amine is soluble in water whereas aniline is not. | | |
| Q.7 | Write reactions to bring about the following reactions : (a) Aniline into p-nitroaniline (b) Aniline into sulphanilic acid | (2) | |
| Q.8 | Write the order of basicity of aliphahtic alkyl amine in gaseous phase. | (2) | |
| Section C Attempt any Two | | | |
| Q.9 | How are diazonium salt react with fluoroboric acid? | (3) | |
| Q.10 | How are amides reduced to amines? | (3) | |
| Q.11 | Explain the influence of solution by water on stabilization of NH_4^+ and aliphatic amines. | (3) | |
| Section D Attempt any One | | | |
| Q.12 | How will you convert aniline into- 1. Benzylalcohol 2. 4–Bromoaniline 3. 1,3,5–Tribromobenzene 4. 2,4,6– Tribromofluorobenzene | (4) | |
| Q.13 | Why the observed order of boiling point of isomeric amines is: Primary amine > Secondary amine > Tertiary amine? | (4) | |